

Novel Self-Regulated Pump (SRP) CO<sub>2</sub> Capture in a CO<sub>2</sub> to Methane system



M. Mahrize<sup>1</sup>, S. Zerbib<sup>1</sup>, N. Shahar<sup>1</sup>, G. Golan<sup>1</sup>  
<sup>1</sup> Ariel University, Science Park, Ariel 40700, Israel, gadygolan@gmail.com



Abstract

Curbing CO<sub>2</sub> emissions in an affordable manner is a global issue. Physical absorption a well-established technology for CO<sub>2</sub> separation from other gases. Physical absorption processes are simple; whereby a gas-liquid contactor saturates the solvent and a flash tank regenerates the solvent. CO<sub>2</sub> is absorbed in the physical solvent in the high-pressure gas liquid contactor and flashed out in the medium and low pressure flash tank. The advantage of using a physical solvent is that the CO<sub>2</sub> is absorbed without any chemical reaction involved, thus it can be flashed out easily by reducing the pressure, passing inert gas through the solvent and mild thermal regeneration. Physical absorption is the best operated at low pressure and low temperature as the solubility of CO<sub>2</sub> in the solvent is high at the particular condition. We are reporting for the first time dynamic physical motion based on a fluid pump designed to transfer fluid to the gas-liquid contactor and create turbulence that significantly increases CO<sub>2</sub> adsorption. Using one pump for this purpose creates small bubbles of CO<sub>2</sub> in the liquid and simultaneously compresses the fluid to create drops of solvent that rise into the gas phase of the reactor. The speed of the pump is regulated by an electrical circuit that measures the percentage of CO<sub>2</sub> adsorption and thus allows the system to be optimized independently until saturation is achieved. Although the results are optimized for physical performance, we assume that Concept has an impact on improving results in each type of CO<sub>2</sub> adsorption.

Innovation of SRP

Nanotechnology today is based on particle size such as powder and their effectiveness in several times of magnitude versus mass or volume. The physical and mathematical affect theory behind it the area of surface that dramatically increases the phenomenon. The bubble and absorption of CO<sub>2</sub> method is very dependent on the viscosity of the liquid which makes it difficult for the bubbles and even reduces the babbble formation only if the pressure that the bubble can overcome the resistance of viscosity in order to create even single bubbles are from time to time. We are present the concept that SRP that part of reactor can to create saturation, mixing and turbulence in the liquid with nan babbles of CO<sub>2</sub> that the physical and chemistry interaction increase exponentially. Additional effect, SRP pumping the liquid that created the effect in the liquid and flow it to spray tower the benefit efficiency of CO<sub>2</sub> absorption with created drops. The other effect is thermal conductivity of drops, bobbles and turbulence that releases heat and does not require a cooling system. The speed of the pump is regulated by an electrical circuit that measures the percentage of CO<sub>2</sub> adsorption and thus allows the system to be optimized independently until saturation is achieved. Preliminary result shown the saturation arrival time and number of cycles are significantly shortened

S.-Y. Lee, S.-J. Park/Journal of Industrial and Engineering Chemistry xxx (2014) xxx-xxx

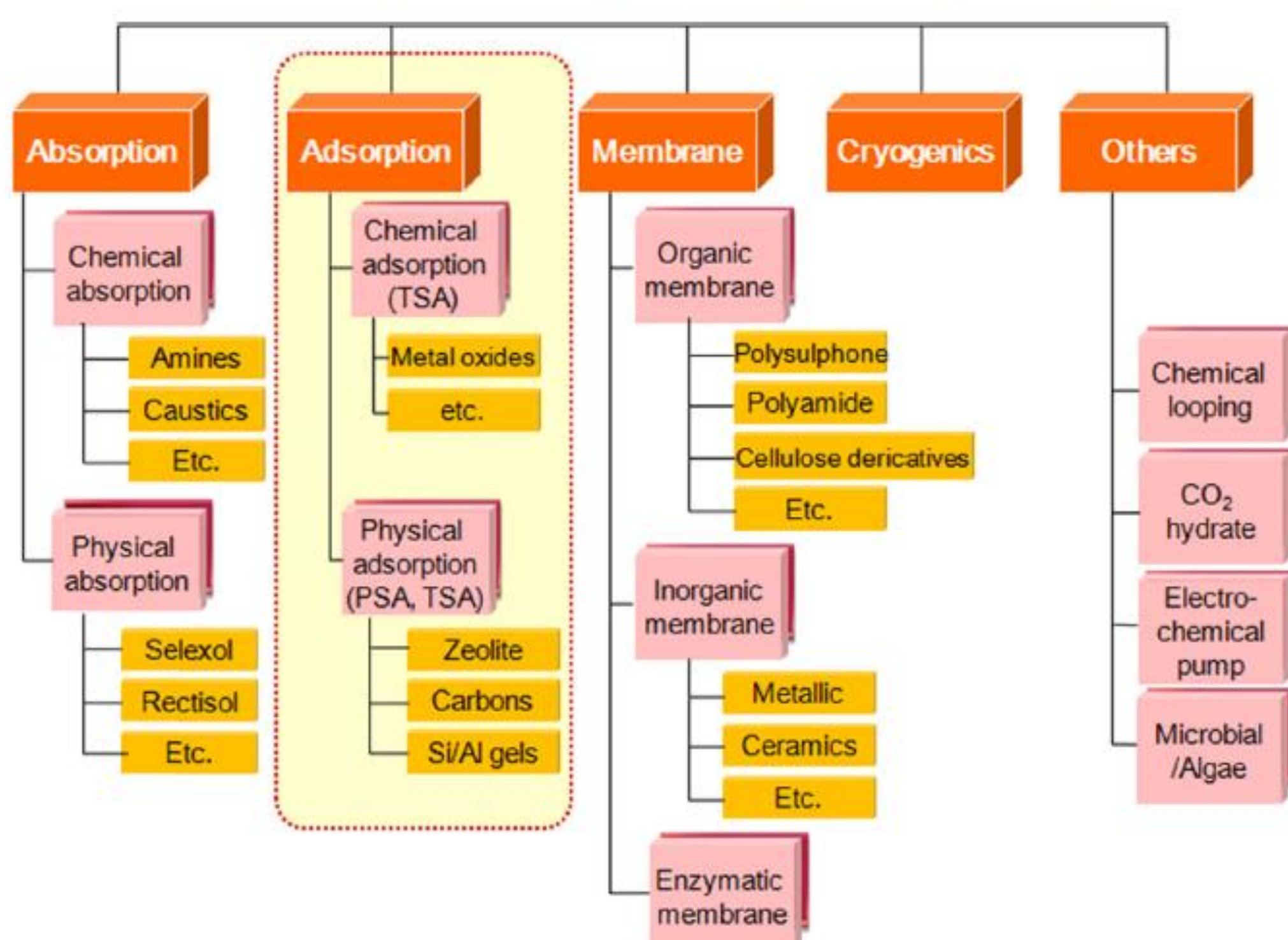


Fig.1 Classification of application technologies for post-combustion capture of CO<sub>2</sub>.



Fig.2 mixing and turbulence by SRP in the MEA solution created nano babbles of CO<sub>2</sub> with high area surface that improve the physical and chemical interaction exponentially